will be announced on

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EV568 EV368 Quercus before semester starts.

Sarah Shujah (librarian)

Biological inorganic chemistry course (CHMD69H3) will bring you the world of inorganic chemistry in living systems. We shall predominantly concentrate on structure and reactivity of metalloproteins: proteins whose structure and/or function depend on the presence of one or more metallic centers; focusing on their structure, reactivity and role in the living systems. Applications of physical methods to the problems in biological inorganic chemistry will also be briefly discussed using specific examples. In order to follow the course material some background in following topics is very important and will be assumed through the course:

3. Metal ion transport and storage

Control of metal ion concentration

Recognition of metal ions

Transport and storage of selected ions

4. Metal ion receptors and signaling

Metalloregulatory proteins

Role of Zn²⁺ binding domains

Role of Ca²⁺ in cells of higher organisms

5. Non-redox metalloenzymes

Overview

Metal dependent lyase and hydrolase

Aconitase

Carboxypeptidase

Carbonic anhydrase

6. Redox metalloproteins.

Electron carriers vs. oxido-reductases

Electron sources and electron chains in living systems

Iron sulfur proteins

Cytochromes

Copper proteins

Respiration

7. Further on oxygen metabolism.

Superoxide dismutase

Peroxidases

- 8. Hydrogen metabolism hydrogenases
- 9. Nitrogen metabolism t nitrogenases

10%
5%
20%
20%
45%

The assignment is going to be posted on Quercus on

and is due in class

You are required to write a paper (1800 to 2000 words in length) and give a 20 min lecture (15 min for your talk + 5 min for Q&A) on a topic you select. The list of suggested topics will be provided separately on the blackboard. You can also suggest a topic that is not on the list but you have to check the suitability of your choice with me. More details on the paper requirements will be provided with the list of possible topics.

To obtain assistance on any assignment.

To provide assistance to another student. This includes showing another student completed

work.

To submit their own work for credit in without the permission of the instructor.

To or any documentation required by the University. This includes, but is not limited to, doctor's notes.

To use or possess an unauthorized aid in any test or exam.

There are other offences covered under the Code, but these are by far the most common. Please respect these rules and the values which they protect. Offences against academic integrity will be dealt with according to the