# CHMC 16 Instrumental Analysis : L Q W H U

Wed 10am-4pmTRACES (EV215) and adjacent lab (EV216)

**Course Description** 

CHMC16 is a laboratory course designed to compliment CHMC11 (Principles of Analytical Instrumentation). This course will provide a practical roduction and experience in the use of modern

The limitation and advantages of thethinique Mixture Analysis by NMR
The use of Electronic References for Quantification

Gas Chromatography and Gas ChromatographyMass Spectrometry (GC-MS)

#### Students learn:

Sample injection methods
Correlation of Theory and Practice through Deemter Plots
Important experimental variables
Calculation of concentrations in unknowns
Quantitative eparations of mixtures
Experimental esign
The limitation and advantages of the technique

Energy Dispersive Xray Fluorescence (EDXRF) Atomic Absorption Spectroscopy (AA) and Total reflection X-ray Fluorescence (TRXRF)

### Students learn:

Compare Various Methods of Metal Analysis
Understand and contrast detection limits
Surface vs. Bulk Analysis
Destructive vs Nordestructive Analysis
Calculation of concentrations in unknowns

Fourier Transform RAMAN (FT -RAMAN)

### Students learn:

Understand the role and potentialFT-RAMAN in a range of analyses Analyze a range of mineral and rock samples Interpretation of Spectral Data Qualification Run libray matching Quantification

### Capillary Electrophoresis (CE)

### Students learn:

Anion and Cation Analysis Reverse and Normal EOF Effects of Buffer, pH, Sample Stacking Indirect UV Detection Developing and programming methods Quantification The coursewill be split into 4 main sections:

Section 1. Three weeks split between MAN, TRXRF, CE

Section 2. Three weeks will be spent learning and applying NMR

Section 3. Three weeks will be spent on GC and MSC

Section 4. Three weeks will be spent learn HPLCMS

### Assessment

There will be no final exam for this course. Students will be assessed on the following criteria.

- 1) 3 x Lab reports. Lab reports are worth 20% each (see later, HPLC no lab report)
- 2) 1 x term paper (25%)
- 3) Ability and Performance the Lab sessions (15%). Remember this is a lab course you will be evaluated on your involvement, safety (lab glasses, coat), you ability to work with your team members, your ideas especially in "the research project section", your ability to keep a lab manual that can both be used to verify your results, and repeat your work, your timeliness, and your ability to organize your time and leave the lab in good shape.

Lab reports are to be written individually and each student will be expected to attach their own copies of the relevant chromatograms, spectra etc with their reports.

Plagiarized reports will not be accepted.

## Late Reports will not be accepted unless you have a Dr's note

Acknowledgement: Some sections of these practicals have modified from "Chemistry Experiments from Instrumental Methods" by Sawyer, Heineman and Beebe.

## Contacts and Office Hrs

Office Hrs: Wed 3.304.30pm. In Environmental NMR Centerake elevator to basement of the science research building and bang on the large double door that are located around the corner)-Enail: andre.simpson@utoronto.ca

# Lab Books, Cleaning Up, and Leaving

### Before Leaving

Make sure all chemicals have been returned and that all apparatus, has been cleaned an returned to its correct location. YOU WILL LOOSE MARKS IF YOU LEAVE A MESS. LOTS OF THE EQUIPMENT YOU WILL BE USING IS VERY EXPENSIVE RESEARCH EQUIPMENT, TREAT IT WITH RESPECT!!

### Lab Manuals:

Lab Manuals are to be kept throughout the course. You must get these initialed by the instructor or demonstrator at the end of each session after you have cleaned up and shut down all the instrumentation properly. Lab manuals be handed in along with the last lab report. It is your responsibility to get you lab book signed each weekF SIGNATURES OR LAB BOOKS ARE MISSING AT THE END OF THE COURSE THEN YOU WILL LOOSE MARKS.

All students are required to make their own notes and observations in the lab books as they feel appropriate

# Example Lab Report + Schedule

# Overall Title (i.e. Gas Chromatography)

Name: Student Number Date:

Names of Other Student in the same practical group

Subtitle (i.e. Week 1: Determination of Optimal Flow Rate in Gas Chromatography)

Make a note here of any special circumstances. For example: "As is was the st week of class only 4 of the 6 compounds were available"

### Treatment of Data and Results

Here you will be expected to work chronologically through the practical write up. You are expected to carry out all the instructions in the "Treatment of DatatiSes" throughout the text. You will be expected to clearly label and hand in all spectra or data collected during the practical. Each individual will be responsible to obtaining his own individual copy of any relevant data. Any appropriate experimental conditions should be clearly marked. In this section discuss any relevant finding or observations that you made during the practical.

### Questions

Answer all questions throughout the text as fully as possible.

### Conclusions

In this section summarize your material findings, and what you think you learnt from the practical session. Address any problems encountered. Suggest improvements that could be made to improve the results.

### Repeat Format for week 2 and hand in complete report.

The practical report should be submitted in full the week following the last practical session on a particular type of instrumentation. Exact dates for handing in each assignment are given on the next page.